

REACTION OF STABLE SULFOXONIUM YLIDES WITH SULFENES<sup>\*1</sup>

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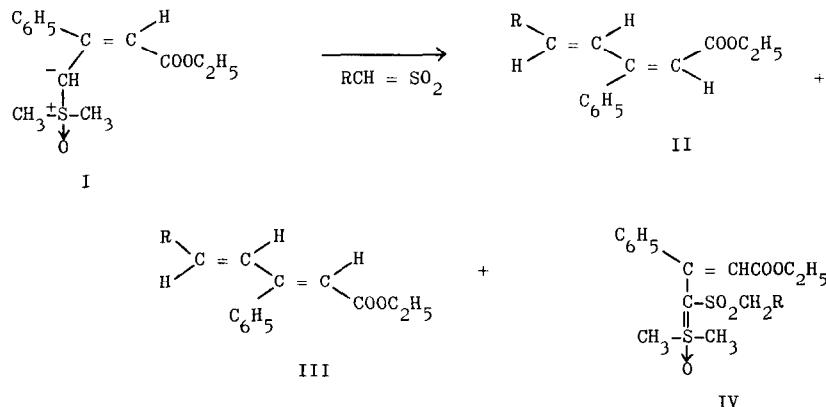
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The appearance of a recent report (1) on the reaction of sulfonium ylides with methane sulfonylchloride derivatives has prompted us to report in a preliminary form our results on a highly stereospecific synthesis of butadiene derivatives by reaction of stable sulfoxonium ylides (2) with sulfenes.

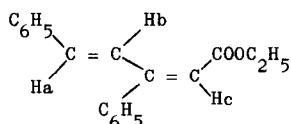
Treatment of sulfoxonium ylide (I)(2a.b.3) with benzylsulfonylchloride in the presence of triethylamine in THF at 4-6° afforded ethyl *trans,trans*-3,5-diphenyl-2,4-pentadienoate (II, R=C<sub>6</sub>H<sub>5</sub>) (63% yield); bp 150-155°/0.0005 mmHg (bath temp.), IR spectrum [ $\nu_{\text{max}}^{\text{Liquid cm}^{-1}}$ : 1709, 1176, 1041, 1027, 990, 980, 873, 775, 757, 706, 696]; UV spectrum [ $\lambda_{\text{max}}^{\text{EtOH m}\mu(\varepsilon)}$ : 230(12,300), 260(9,900), 316(26,800)], *trans,cis*-isomer (III, R=C<sub>6</sub>H<sub>5</sub>) (2.3% yield) and C-sulfonylated product (IV, R=C<sub>6</sub>H<sub>5</sub>) (1.3% yield), mp 157-158°, IR spectrum [ $\nu_{\text{max}}^{\text{Nujol cm}^{-1}}$ : 1709, 1468, 1383, 1297, 1170, 1111, 1042], UV spectrum [ $\lambda_{\text{max}}^{\text{EtOH m}\mu(\varepsilon)}$ : 264(11,500), 364(2,550)]. These NMR spectra are shown in Table I and III.



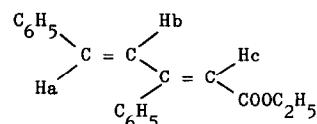
\*1 This paper forms part V of a series entitled "Sulfoxonium Ylide Chemistry."

For previous papers, See Ref. 2.

TABLE I



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III

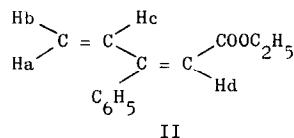
$R=C_6H_5$	$CH_3$	$O-CH_2Me$	Ha	Hb	Hc	Aromatic protons
II	8.68(t) . . . 5.74(q)		3.37(q)	1.42(q)	4.17(q)	2.6 (m)
	$J = 7$ cps		$J_{ab}=16$ cps	$J_{ac}=J_{bc}=1$ cps		
III	8.94(t)	5.99(q)	3.62(q)	2.82(q)	3.89(q)	2.6 (m)
	$J = 7$ cps		$J_{ab}=16$ cps	$J_{ac}=J_{bc}=1$ cps		

The signals of the proton H<sub>b</sub> in the trans,trans-isomer (II, R=C<sub>6</sub>H<sub>5</sub>) appear at an abnormally low field, which is ascribed to the anisotropic effect of carboethoxy group located on the same side of the diene system (4).

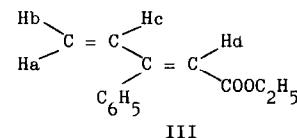
In the case of the reaction of methane sulfonylchloride with the ylide (I), the corresponding dienes, ethyl *trans*-3-phenyl-2,4-pentadienoate (II, R=H); bp 90–95°/0.03 mmHg (bath temp.) (54% yield), IR spectrum [ $\nu_{\text{max}}^{\text{Liquid cm}^{-1}}$ : 1709, 1170, 1000, 934, 870, 775, 699], UV spectrum [ $\lambda_{\text{max}}^{\text{EtOH m}\mu(\varepsilon)}$ : 234(plateau)(9,160), 262(13,750)], the *cis*-isomer (III, R=H) (1.2% yield) and a C-sulfonylated product (IV, R=H), mp 155–156° (10.6% yield) were obtained. These NMR spectra were shown in Table II and III.

TABLE II

### NMR Spectra of II and III ( $R=H$ )



II



III

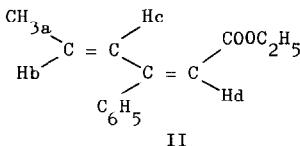
R=H	CH <sub>3</sub>	O-CH <sub>2</sub> -Me	Hα	Hβ	Hγ	Hδ	Aromatic protons
II	8.70(t)	5.76(q)	4.69(octet)	4.39(sextet)	2.02(octet)	4.16(quintet)	2.63(s)
	J=7 cps	Jab=Jbd=2 cps	Jac=17 cps	Jbc=11 cps	Jad=Jcd=1 cps		
III	8.96(t)	5.99(q)	4.91(sextet)	4.53(sextet)	3.32(octet)	3.99(q)	2.6(m)
	J=7 cps	Jab=Jbd=Jad=Jcd=1 cps	Jac=17 cps	Jbc=10 cps			

TABLE III  
NMR Spectra of IV

IV R	$\text{CH}_3$	$\text{O}-\text{CH}_2-\text{Me}$	$\text{CH}_3-\overset{\text{O}}{\underset{\text{S}}{\text{---}}} \text{---CH}_3$	$\text{SO}_2\text{CH}_3$	$\text{SO}_2\text{CH}_2\text{C}_6\text{H}_5$	$-\text{C}=\text{CH}-$	Aromatic protons
$\text{C}_6\text{H}_5$	8.67(t) $J=7$ cps	5.75(q) $J=7$ cps	6.47(s)	-	5.87(s)	3.55(s)	2.4(m)
H	8.90(t) $J=7$ cps	5.96(q)	6.49(s)	7.10(s)	-	3.78(s)	2.6(s)

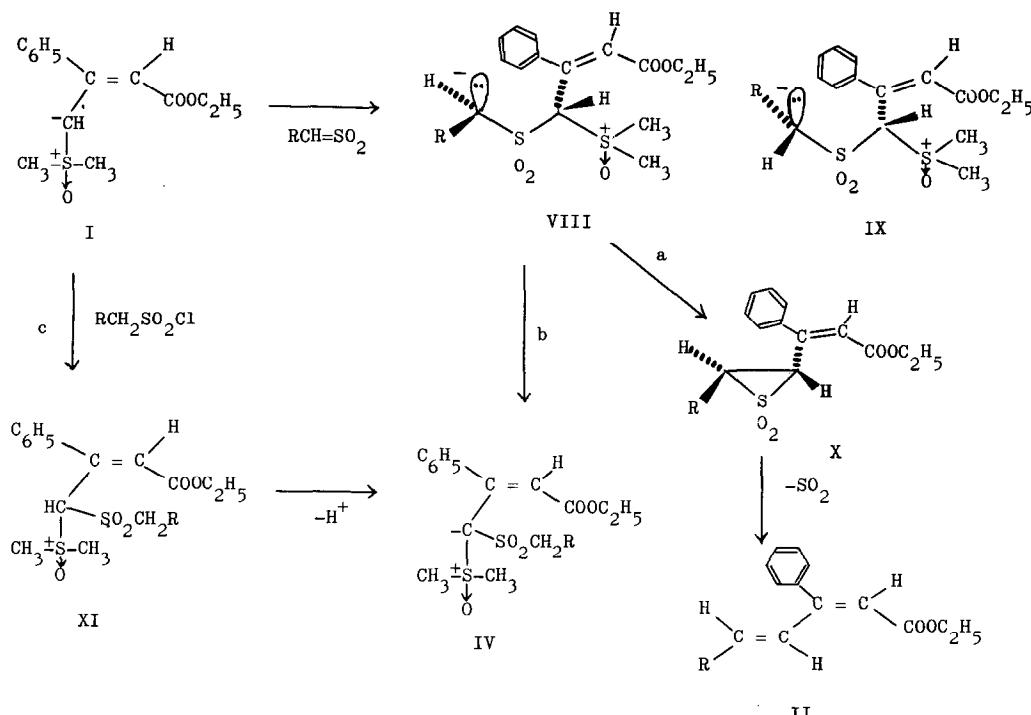
Similarly the reaction of the ylide (I) with ethane sulfonylchloride gave trans,trans-diene (II, R=CH<sub>3</sub>); bp 100–105°/0.02 Hg (bath temp.) (40.8% yield), IR spectrum [Liquid  $\nu_{\text{max}}^{\text{cm}^{-1}}$ : 1712, 1163, 980, 873, 775, 706], UV spectrum [ $\lambda_{\text{max}}^{\text{EtOH}} \text{ m}\mu(\epsilon)$ : 231(8,200), 274(17,240)].

TABLE IV  
NMR Spectra of II



R=CH <sub>3</sub>	$\text{CH}_3$	$\text{O}-\text{CH}_2-\text{Me}$	$\text{CH}_3\text{a}$	Hb	Hc	Hd	Aromatic protons
8.69(t)	5.77(q)	8.13(octet)	4.14(octet)	2.24(octet)	4.32(q)	2.65(s)	
$\text{Jab}=7\text{cps} \quad \text{Jac}=2\text{cps} \quad \text{Jbc}=16\text{cps} \quad \text{Jad}=\text{Jbd}=\text{Jcd}=1 \text{ cps}$							

The highly stereospecific formation of double bond would proceed through a thermal cis-elimination (5) of sulfur dioxide from a trans-episulfone (X). The addition of the ylide-carbanion to sulfenes would form the transition states (VIII and IX). The large cinnamyl and R (R=CH<sub>3</sub>, C<sub>6</sub>H<sub>5</sub>) groups interfere each other in IX but not in VIII. Thus the elimination of DMSO proceeding through transition state (VIII) takes place more readily than that proceeding through IX. The C-sulfonylated product would be formed either by proton transfer of VIII or IX (path b), or by a direct substitution of the sulfonylchloride with ylides followed by stabilization to a C-sulfonylated ylide (IV) (path c). We are currently extending this work to the reaction of other ylides with sulfenes in the hope of "a new type of Wittig reaction" with high stereospecificity.



## References

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